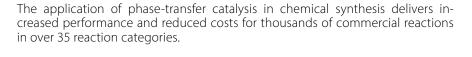


## Phase-Transfer Catalysts

Aliquat ®





- · Increase yield
- Improve selectivity
- Reduce reaction time and temperature
- Reduce excess reactants and waste
- · Replace expensive hazardous strong bases with NaOH
- Replace undesired solvents such as DMSO, NMP, DMF
- · Enhance process safety

#### **Products overview:**

<b>Product name</b>	Chemical description
Aliquat 336	Tricaprylylmethylammonium chloride

## Phase-transfer catalysis in a nutshell

For a desired reaction R-Y -> R-X, the reagent MX might be dissolved in the aqueous phase whereas the substrate is in the organic phase. This usually results in poor conversion.

Phase-transfer catalysis is based on the ability of a catalyst (Q+), typically a quaternary ammonium cation, to transfer and actually solubilize an otherwise hydrophilic anion in an organic phase. The anion (X-) is typically a nucleophile, a base, an oxidizing or a reducing agent. Once in the organic phase, the anion can react with enhanced reactivity with an organic soluble substrate (R-Y).







# Phase-Transfer Catalysts

Aliquat ®





Etherification

C - Alkylation

N - Alkylation

S - Alkylation

Dehydrohalogenation

Condensations

Aldol

Michael Wittig

Darzens

Chiral Alkylation & Addition

Carbenes

Deuteration

## **Nucleophilic Aliphatic Substitution**

Esterification

Transesterification

CN-

F-

Br-

|-

N<sub>3</sub>-

SČN-

OCN-

SO<sub>2</sub><sup>2</sup>-

NO<sub>2</sub>-, NO<sub>3</sub>-

#### Oxidation

Epoxidation

H,O,

CĺO-, BrO-

Oxygen / Air

 $MnO_4^{-}, Cr_2^{O_7^{2}}$ 

S<sub>2</sub>O<sub>8</sub><sup>2</sup>

Óxone

Superoxide

Electrochemistry

#### Polymerization

Condensation

Radical

### **Strong Acid Reactions**

HCl, HBr Additions

Chlormethylation

Ether cleavage

#### **Transition Metal Co-Catalysis**

Reduction

Carbonyation

Coupling reactions

#### Reduction

BH<sub>4</sub>-

Hydrogenation

S<sub>2</sub>O<sub>4</sub><sup>2</sup>-

Aliquat <sup>®</sup> is a brand name of the company BASF.



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